

Brian McBurney,<sup>a</sup>  
Guy Crundwell,<sup>a</sup>  
James B. Updegraff III,<sup>b,\*</sup>  
Matthias Zeller<sup>b</sup> and  
Allen D. Hunter<sup>b</sup>

<sup>a</sup>Department of Chemistry, Central Connecticut State University, New Britain, CT 06053, USA, and <sup>b</sup>Department of Chemistry, Youngstown State University, One University Plaza, Youngstown, Ohio 44555-3663, USA

Correspondence e-mail:  
crundwellg@mail.ccsu.edu

#### Key indicators

Single-crystal X-ray study  
 $T = 100$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å  
 $R$  factor = 0.027  
 $wR$  factor = 0.075  
Data-to-parameter ratio = 9.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## 5-Nitrothiophene-2-carbaldehyde

In the crystal structure of the title compound,  $\text{C}_5\text{H}_3\text{NO}_3\text{S}$ , determined at 100 K, the nitro group and the aldehyde group make angles of 8.37 (16) and 8.5 (4)°, respectively, with the plane of the thiophene ring.

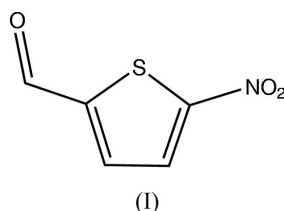
Received 28 February 2005

Accepted 30 March 2005

Online 9 April 2005

#### Comment

In the title compound, (I), all bond distances and angles fall within the normal ranges. The molecule is essentially planar; however, the nitro group and the aldehyde group are at angles of 8.37 (16) and of 8.5 (4)°, respectively, with plane of the thiophene ring. In the crystal structure, molecules are connected by weak  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds (see Table 1).



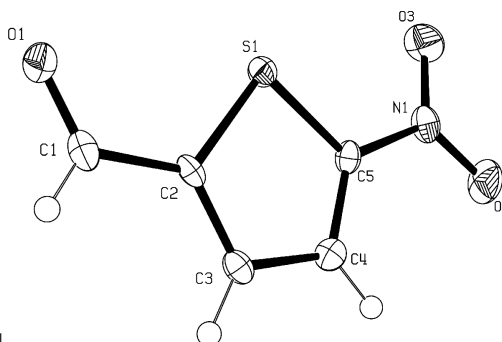
#### Experimental

The title compound was purchased from Aldrich (98% purity) and was recrystallized from ethanol prior to its use as a reactant in the Adler synthesis for porphyrins (Adler *et al.*, 1967). White crystals of the title compound were grown from the slow evaporation of ethanol solutions at 278 K.

#### Crystal data

$\text{C}_5\text{H}_3\text{NO}_3\text{S}$   
 $M_r = 157.14$   
Orthorhombic,  $Pna2_1$   
 $a = 11.4180$  (7) Å  
 $b = 13.9536$  (8) Å  
 $c = 3.8525$  (2) Å  
 $V = 613.79$  (6) Å<sup>3</sup>  
 $Z = 4$   
 $D_x = 1.701$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation  
Cell parameters from 5184 reflections  
 $\theta = 2.3$ – $28.3^\circ$   
 $\mu = 0.46$  mm<sup>-1</sup>  
 $T = 100$  (2) K  
Needle, white  
 $0.50 \times 0.05 \times 0.05$  mm



**Figure 1**

A view of (I) (Farrugia, 1997). Displacement ellipsoids are drawn at the 50% probability level.

Data collection

Bruker SMART APEX diffractometer  
 $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Sheldrick, 2003)  
 $T_{\min} = 0.921$ ,  $T_{\max} = 0.980$   
 5572 measured reflections

872 independent reflections  
 871 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.025$   
 $\theta_{\text{max}} = 28.3^\circ$   
 $h = -15 \rightarrow 15$   
 $k = -18 \rightarrow 18$   
 $l = -5 \rightarrow 5$

Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.027$   
 $wR(F^2) = 0.075$   
 $S = 1.18$   
 872 reflections  
 95 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0443P)^2 + 0.2249P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.001$   
 $\Delta\rho_{\text{max}} = 0.37 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.19 \text{ e } \text{\AA}^{-3}$

Table 1

Hydrogen-bonding geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$C1-H1 \cdots O1^i$	0.93 (3)	2.52 (3)	3.373 (3)	152 (2)
$C4-H4A \cdots O1^{ii}$	0.93	2.46	3.364 (3)	163

Symmetry codes: (i)  $1-x, -y, z - \frac{1}{2}$ ; (ii)  $x - \frac{1}{2}, \frac{1}{2} - y, z - 1$ .

Thienyl H atoms were included in calculated positions with a C–H distance of 0.95  $\text{\AA}$  and were included in the refinement in riding-model approximation with  $U_{\text{iso}} = 1.2U_{\text{eq}}$  of the carrier atom. The aldehyde H atom was found in a difference map during the initial

stages of refinement and was refined independently with an isotropic displacement parameter. Refinements to determine the absolute structure gave inconclusive results [yielding a Flack (1983) parameter of 0.15 (9)] and the Friedel equivalents were therefore merged before the final refinement cycles.

Data collection: SMART (Bruker, 1999); cell refinement: SAINT-Plus (Bruker, 1999); data reduction: SAINT-Plus; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: SHELXL97.

GC acknowledge the Donors of the American Chemical Society Petroleum Research Fund for primary support of this research (No. 38867-B5m). This research was also funded in part by CCSU-AAUP research grants and CCSU Faculty–Student Research Grants. MZ and JU were supported by NSF grant No. 0111511, and the diffractometer was funded by NSF grant No. 0087210, by the Ohio Board of Regents grant No. CAP-491, and by YSU.

References

Adler, A. D., Longo, F. R. & Goldmacher, J. (1967). *J. Org. Chem.* **32**, 476.  
 Bruker (1999). SMART and SAINT-Plus. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.  
 Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.  
 Sheldrick, G. M. (2003). SADABS. Version 2.10. University of Göttingen, Germany.  
 Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.